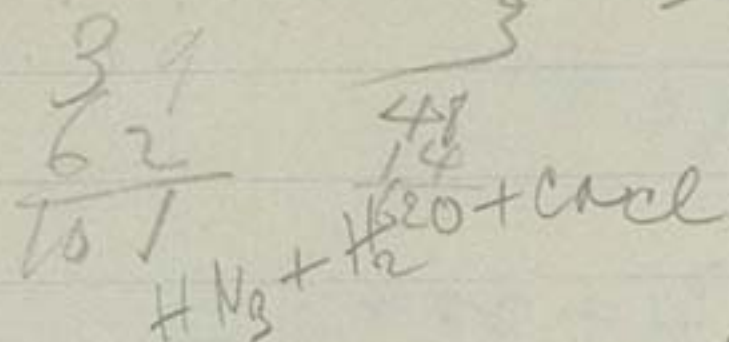


$$\begin{array}{r} 59 \\ 14 \\ \hline 48 \\ \hline 101 \\ \hline \end{array} \quad \begin{array}{r} 15 \\ 3 \\ \hline 48 \\ \hline 14 \\ \hline 19 \end{array}$$



$$\begin{array}{r} 2 \\ 32 \\ 64 \\ \hline 98 \end{array} \quad \begin{array}{r} 32 \\ 4 \\ \hline 36 \end{array}$$

98:63::1:1

$$\begin{array}{r} 63 \times 20 \\ 91 \\ 16 \\ \hline 34 \\ 64 \\ \hline 98 \end{array} \quad \begin{array}{r} 11 \\ 48 \\ \hline 259 \\ 39 \\ 64 \\ \hline 202105 \\ 98 \end{array}$$

Chemistry
2nd Prep Class.

98:101::1:202

$$\begin{array}{r} 144 \\ 32 \\ \hline 176 \\ 34 \\ \hline 210 \end{array}$$

1. How is Ammonia produced both in nature, and artificially? Give some of its chief properties.
2. State the chief properties of Chlorine, & compare them with those of Bromine & Iodine.
3. State the preparation & the properties of Hydrochloric acid.
4. How many grams of nitric acid can be obtained from 202 grams of Nitre? Also calculate the amount of sulphuric acid required.

$$\begin{array}{r} 17 \\ 14 \\ \hline 31 \end{array}$$

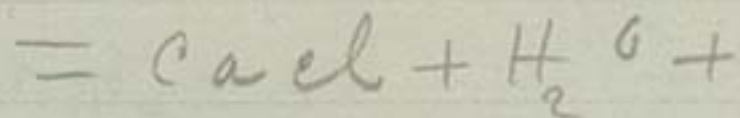
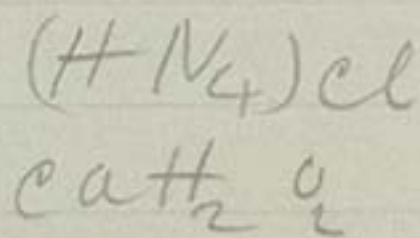
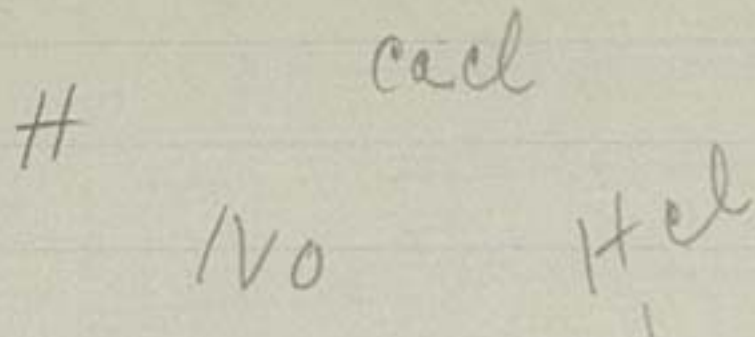
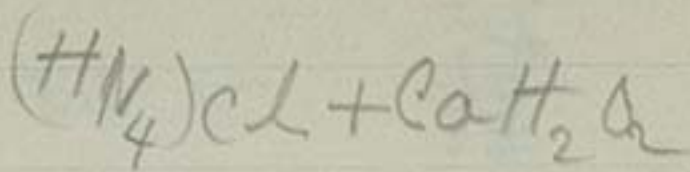
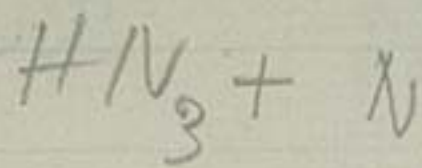
$$\begin{array}{r} 15 \\ 48 \\ \hline 63 \end{array}$$

39 K

$$\begin{array}{r} 32 \\ 64 \\ \hline 96 \end{array}$$

$$\begin{array}{r} 16 \\ 4 \\ \hline 20 \end{array}$$

$$\begin{array}{r} 39 \\ 14 \\ \hline 53 \\ 48 \\ \hline 101 \end{array}$$



202
98
1.61.6
1818
19796 | 501
189
719

89

